(T(5th Sm.)-Chemistry-H/CC-11/CBCS)

2020

CHEMISTRY — HONOURS

Paper : CC–11

(Physical Chemistry - 4)

Full Marks : 50

The figures in the margin indicate full marks.

Candidates are required to give their answers in their own words as far as practicable.

Answer question no. 1 and any eight questions of the rest.

1. Answer any ten of the following :

1×10

(a) Write down the Schrödinger equation for the one-dimensional simple harmonic oscillator (SHO).

(b) Show that the SHO wave function
$$\psi_0(x) = \left(\frac{\alpha}{\pi}\right)^{\frac{1}{4}} e^{-\frac{\alpha x^2}{2}}$$
 is normalized.

- (c) Write down the form of the \hat{L}_z in terms of $\hat{x}, \hat{y}, \hat{z}$ and $\hat{p}_x, \hat{p}_y, \hat{p}_z$.
- (d) Write down the energy expression for the rigid rotator explaining the terms and mention the admissible values of the rotational quantum number J.
- (e) For the H-atom write down the *normalized* ϕ -function $\Phi(\phi)$, mentioning the restriction on *m*.
- (f) Write down the normalized radial wave function for the 1s orbital of the hydrogen atom, $R_{10}(r)$, mentioning the relevant term used.
- (g) State the Born-Oppenheimer approximation qualitatively.
- (h) Write the expression of the Hamiltonian operator of Lithium atom.
- (i) Suppose 10 data points for each of x and f(x) are given. Write down the strategy for integrating f(x) using a combination of Simpson's one-third rule and Trapezoidal rule.
- (j) For what order of polynomials does the Trapezoidal Rule furnish an *exact* value of the integral? Give one example.
- (k) Write down the relationship between the partition function of a statistical system and its entropy.
- (1) What is the dimension of the parameter β , as used in the Boltzmann distribution function?
- (m) Write down the equation for thermodynamic probability of a state having a total number of particles N distributed among the different energy levels ϵ_i , with n_i in the respective energy levels.
- (n) Name the *three* constant parameters characteristic of the elements of a Grand Canonical Ensemble.
- (o) What should be the degeneracy of the ground state for a statistical system so that it exhibits 'residual entropy'?

T(5th Sm.)-Chemistry-H/CC-11/CBCS

2. (a) In the mathematical form of the Boltzmann distribution the sign before β is negative. Justify.

(2)

- (b) Deduce the 1st-derivative Central Difference formula in connection with numerical differentiation, using the Taylor Series expansion. 2+3
- 3. Set up the Hamiltonian operator for H_2^{\oplus} (hydrogen molecule ion) in atomic units. The LCAO MO may be written as $\psi_{\pm} = C_1 1 S_A \pm C_2 1 S_B$ (terms have their usual meaning). Draw the plots of ψ_{\oplus}^2 and ψ_{\odot}^2 against the distance of nuclear separation (R) and hence label ψ_{\oplus} and ψ_{\ominus} as bonding or anti-bonding molecular orbital. Give arguments for choosing $C_1 = C_2 = C$.
- 4. (a) Calculate the positive root of the equation $x^2 + 2x 2 = 0$. Correct upto two significant figures by the Newton-Raphson method.
 - (b) Show that $\left[\hat{L}^2, \hat{L}_z\right] = 0$, where the terms have got their usual significance. 3+2
- 5. The ground state SHO wave function is given by $\psi_0(x) = (\alpha/\pi)^{\frac{1}{4}} \exp(-\alpha x^2/2)$. Find $\langle x \rangle$, $\langle x^2 \rangle$,

 $\langle p_x \rangle, \langle p_x^2 \rangle$. Hence show that the Uncertainty Principle of Heisenberg holds true for this eigenstate.

Given:
$$\int_{0}^{\infty} x^{2n} \exp\left(-ax^{2}\right) dx = \frac{1 \cdot 3 \dots \cdot (2n-1)}{2^{n+1}} \sqrt{\frac{\pi}{a^{2n+1}}}$$
$$\int_{0}^{\infty} x^{2n+1} \exp\left(-ax^{2}\right) dx = \frac{n!}{2a^{n+1}}$$

6. (a) Consider the following table :

x	1.0	2.0	3.0	4.0	5.0	6.0	7.0
у	2.0	5.0	7.0	10.0	12.0	15.0	19.0

Calculate the slope and the intercept of the best fit straight line using these points.

- (b) A certain molecule has a non-degenerate excited state lying 540cm⁻¹ above the non-degenerate ground state. At what temperature is the population of the excited state just 10% of the ground state?
 3+2
- 7. (a) Calculate the difference in the barometric reading between the tenth floor and ground floor of a building where each floor is 3 meter high. Express your answer in *torr*. [Given $T = 27^{\circ}$ C and air is considered as a mixture of N₂ and O₂ in the mole-ratio 4 : 1, N = 14, O = 16, at. wt.]

[T(5th Sm.)-Chemistry-H/CC-11/CBCS]

(b) Given:
$$\psi_{2p+1} = \frac{1}{8\sqrt{\pi}} \left(\frac{1}{a}\right)^{5/2} r e^{-r/2a} \cdot \sin \theta \cdot e^{i\phi}$$

 $\psi_{2p-1} = \frac{1}{8\sqrt{\pi}} \left(\frac{1}{a}\right)^{5/2} r e^{-r/2a} \cdot \sin \theta \cdot e^{-i\phi}$

(for hydrogen atoms)

Find the real wave functions by taking two suitable combinations of the above two functions.

3+2

- 8. (a) With the help of a suitably labelled diagram, briefly explain the principle of 'Adiabatic-Demagnetization'.
 - (b) From Boltzmann distribution, show that if the energy of the *three* consecutive levels are in arithmetic progression, the number of particles in the respective energy levels will be in geometric progression. 3+2
- 9. (a) From the Boltzmann distribution function, prove that $U = -RT^2 \frac{d \ln q}{dT}$, terms have their usual significance.
 - (b) Taking trial function, $\psi = e^{-cx^2}$ for the ground state of one-dimensional harmonic oscillator, find the optimum value of the parameter 'c' for the ground state, using the Variation Method.

Given: (i)
$$\int_{0}^{\infty} x^{n} e^{-qx} dx = \frac{n!}{q^{n+1}}, \quad n > -1, \quad q > 0;$$
 (ii) $\int_{0}^{\infty} e^{-bx^{2}} dx = \frac{1}{2} \left(\frac{\pi}{b}\right)^{\frac{1}{2}}, \quad b > 0$
2+3

- 10. (a) Write down the LCAO MO wave function for H_2 molecule. Comment on its limitation in describing the dissociation limit. Suggest an improvement with the VB treatment.
 - (b) State Planck's formulation of the Third law of Thermodynamics and explain the same. 3+2
- 11. The spherical harmonic $Y_{+1}^{-1}(\theta,\phi)$ is given as $(3/8\pi)^{1/2} \sin \theta e^{-i\phi}$. Show that Y_{+1}^{-1} is normalized and also that it is orthogonal to Y_0^0 . Make use of de Moivre's Theorem in the latter case.
- 12. (a) State and explain the Nernst Heat Theorem.
 - (b) Write down the Schrödinger equation for the rigid rotator in three dimensions in spherical polar coordinates. Separate the variables. 2+3

Please Turn Over

(3)

- 13. (a) Consider a system 'A' consisting of subsystems 'A₁' and 'A₂' for which $W_1 = 1 \times 10^{20}$ and $W_2 = 2 \times 10^{20}$.
 - (i) What is the numerical configuration available to the combined system?
 - (ii) Compute the entropies S, S_1 and S_2 .
 - (iii) What is the significance of this result?
 - (b) What do you understand by the classical turning point of an one-dimensional harmonic oscillator?

3+2