T(1st Sm.)-Chemistry-H/Pr./CC-2P/CBCS

# 2020

### CHEMISTRY — HONOURS — PRACTICAL

### Paper : CC-2P

#### Full Marks : 30

The figures in the margin indicate full marks.

#### **Physical Chemistry**

#### (Marks : 20)

- 1. Determine the Viscosity Coefficient of a given solution using Ostwald Viscometer.
  - (a) Write down the theory using the following points :
    - (i) Newton's law of Viscosity, viscosity coefficient and it's unit.
    - (ii) Poiseuille's Equation and explanation of the terms.
    - (iii) Draw a neat diagram of the Ostwald Viscometer to show the pressure differences between the two arms of the viscometer.
    - (iv) Derivation of the Working Formula.
  - (b) Determine the ratio of viscosity coefficient of the supplied solution to that of water at the experimental temperature. Given : time of flow for water = 80 sec, time of flow for the solution of same volume through the same viscometer = 100 sec. Density of Water at the experimental temperature =  $1.00 \text{ g.cm}^{-3}$ . Density of the supplied solution at experimental temperature =  $1.12 \text{ g.cm}^{-3}$ .

## Organic Chemistry (1B)

#### (Marks : 10)

2. Draw a neat diagram of boiling point apparatus and properly label its different components.

	(a) Diagram	
	(b) Proper labelling.	2+2
3.	Suggest which among the following pairs would have a higher boiling point? Explain why.	
	(a) <i>n</i> -butyl alcohol and isobutyl alcohol.	

(b) Cyclohexanol and ethyl methyl ketone.

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